TUNTWIN's Workshop

Session A: Introductory session







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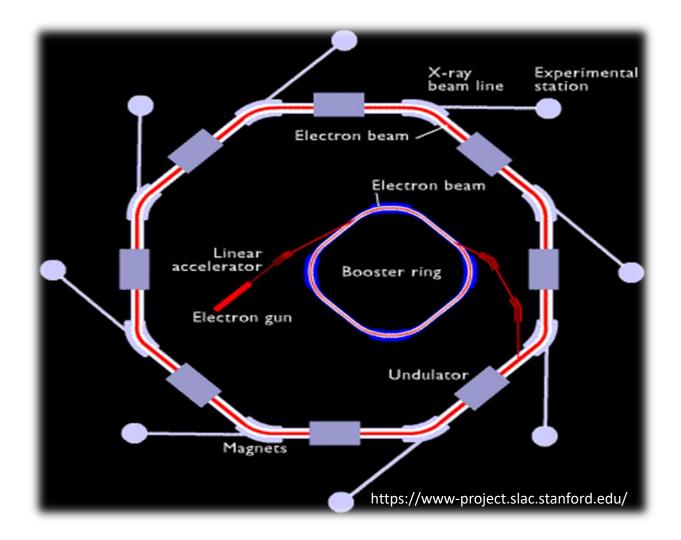
Session A: Introductory session

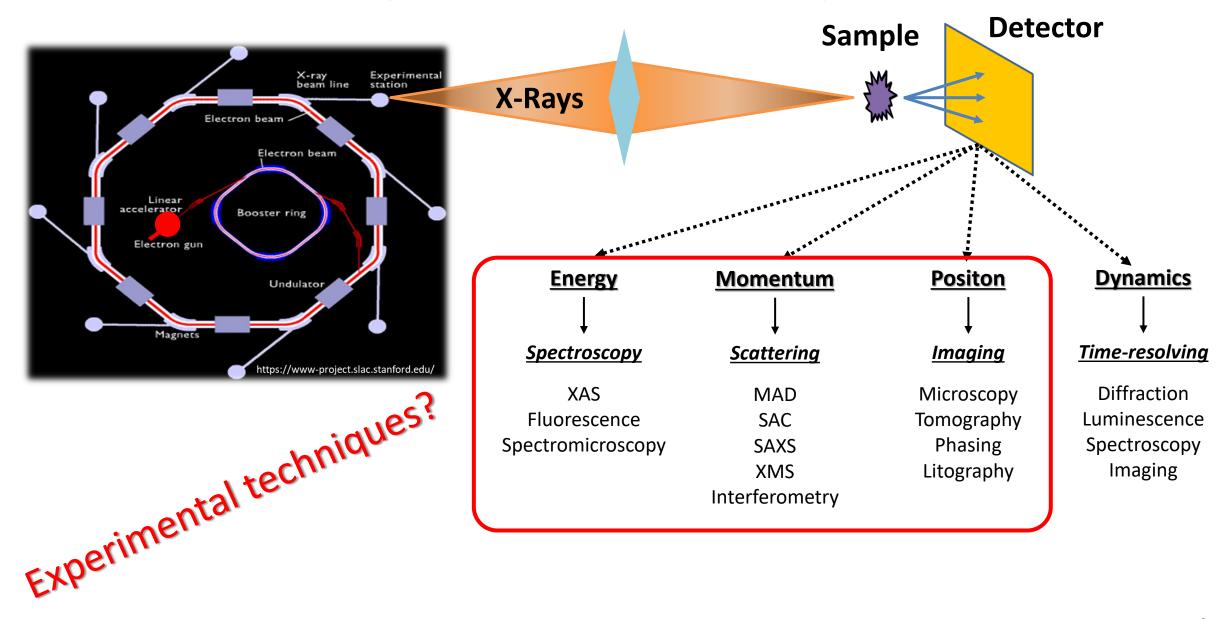
An overview of the synchrotron radiation techniques: spectroscopy, scattering and imaging

Iris H.Valido

- Brief Introduction
- Spectroscopy
- Scattering

Imaging





An overview of the synchrotron radiation techniques: Brief introduction

An overview of the synchrotron radiation techniques: Brief introduction

Experimental techniques

<u>Spectroscopy</u>	01. Low-Energy Spectroscopy 02. Soft X-Ray Spectroscopy 03. Hard X-Ray Spectroscopy	
	04. Optics/Calibration/Metrology	



Chemical Information

<u>Scattering</u>	 05. Hard X-Ray Diffraction 06. Macromolecular Crystallography 07. Hard X-Ray Scattering 08. Soft X-Ray Scattering
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09. Hard X-Ray Imaging





Spatial distribution

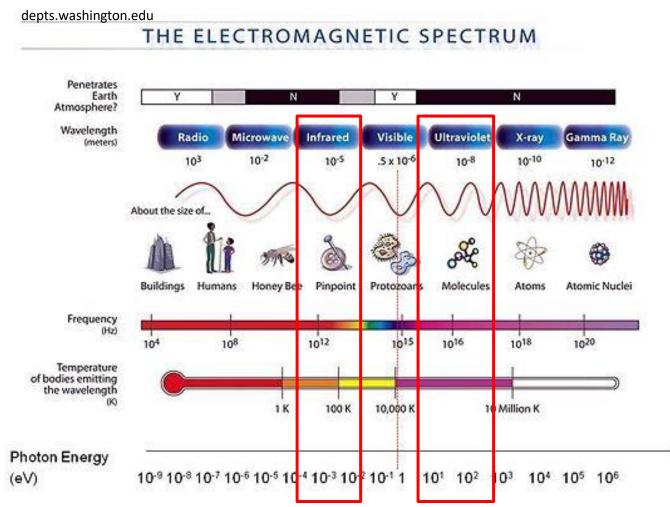
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10. Soft X-Ray Imaging 11. Infrared Imaging 12. Lithography

<u>Spectroscopy</u>

01. Low-Energy Spectroscopy02. Soft X-Ray Spectroscopy03. Hard X-Ray Spectroscopy04. Optics/Calibration/Metrology

Spectroscopy: 01. Low-Energy Spectroscopy



Radiation:

- UV region of the spectrum: 10-100 eV
- IR and Terahertz region: < 1 eV

How it works:

The sample is illuminated with light (in the different spectral region, depending on the technique) and the product particles, which can be photons, electros, ions, atomic clusters, etc. are detected and analyzed.

Spectroscopy: 01. Low-Energy Spectroscopy

Purpose:

- **UV**: elucidation of bonding in solids, surfaces and molecules; investigation of electron-electron correlations in solids, atoms and ions; and studying reaction pathways in chemical dynamics
- **IR**: study the activation of vibrational and other modes of excitement of atomic bonds.
- **Terahertz radiation**: since it is non-ionizing, it does not damage tissues, which makes it useful for the study of biological systems.

Typical applications:

- Complex materials
- Surfaces, clusters
- Atomic & molecular physics, astrophysics
- Combustion, chemical dynamics
- Biological systems

Spectroscopy: 01. Low-Energy Spectroscopy

Includes:

- Vacuum-ultraviolet (VUV) spectroscopy
- Photoelectron spectroscopy (PES)
- Angle-resolved photoelectron spectroscopy (ARPES)
- Photon-ion spectroscopy
- Infrared (IR) spectroscopy
- Terahertz (THz) spectroscopy
- Ultraviolet photoemission spectroscopy (UPS)
- Cold-targed recoil-ion momentum spectroscopy (COLTRIMS)
- Photoelectron-photoion coincidence (PEPICO) spectroscopy
- Fourier transform infrared (FTIR) spectroscopy

Why synchrotron?:

- High signal to noise (S/N) ratio with very high resolving power
- Its high brightness and possibility to achieve small spot sizes for analysis

Spectroscopy: 02. Soft X-Ray Spectroscopy

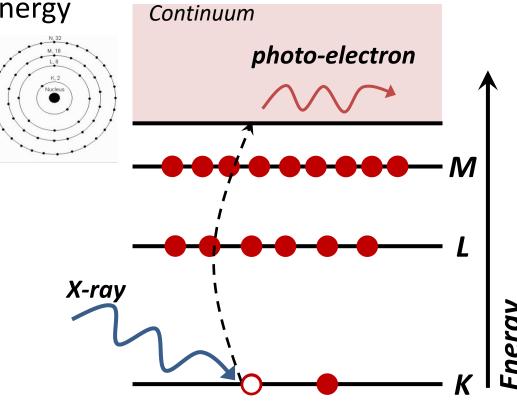
How it works:

In a basic and summarize level, the absorption, transmission, or reflectivity of a sample is measured as a function of photon energy *Continuum*

An atom will absorb an X-ray when the energy is transferred to a core-level electron, and it is ejected from the atom.

The atom is then excited with an empty electronic level (core hole).

The excess of energy form the X-ray is given to the ejected photo-electron



Spectroscopy: 02. Soft X-Ray Spectroscopy

Purpose:

Soft X-Ray Spectroscopies employ the excitation of electrons in relatively shallow core levels (100-2000 eV) to probe the electronic structure of various kinds of matter

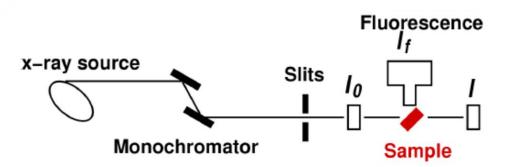
Typical applications:

- Complex materials
- Magnetic materials
- Environmental science, wet samples at ambient pressure
- Catalysis

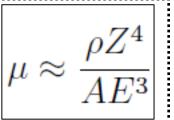
- X-ray absorption spectroscopy (XAS)
- Near-edge x-ray absorption fine structure (NEXAFS) spectroscopy
- X-ray magnetic circular dichroism (XMCD)
- Soft X-ray emission spectroscopy (SXES)
- X-ray photoemission spectroscopy (XPS)
- Auger spectroscopy

Spectroscopy: 03. Hard X-Ray Spectroscopy

How it works:



μ depends strongly on x-ray energy E atomic number Z, and also on density ρ and Atomic mass A:



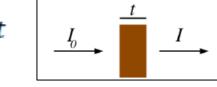
• **Transmission:** measure what is transmitted through the sample: $I = I_0 e^{-\mu(E)t}$

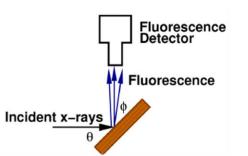
Appropriate for concentration samples ≥ 10 wt.%.

• Fluorescence: measure fluorescent x-rays from the re-filling of the core hole: $\mu(E) \propto I_f/I_0$

Appropriate for dilute elements \leq 2 wt.%.

We need a measurement of $\mu(E)$ to ~0.1%, but with an energy tunable x-ray source, the measurements are fairly easy





Spectroscopy: 03. Hard X-Ray Spectroscopy

Purpose:

Investigate geometric and electronic structures. The method is element-, oxidationstate, and symmetry-specific

Typical applications:

- Characterization of new and promising materials
- Elucidation of dilute chemical species of environmental concern

- X-ray absorption spectroscopy (XAS)
 - Extended X-ray absorption fine structure (EXAFS) spectroscopy
 - X-ray absorption near-edge structure (XANES) spectroscopy
- Near-edge x-ray absorption fine structure (NEXAFS) spectroscopy
- X-ray magnetic circular dichroism (XMCD)

Spectroscopy: 04. Optics/Calibration/Metrology

Purpose:

Calibration of optical components used for X-ray detection, imaging, and spectroscopy as diagnostics. It is important for synchrotron radiation facilities to have some beamline(s) dedicated to the testing and calibration of optics and detectors for use on other beamlines

Typical applications:

- Nuclear physics (fusion plasma) diagnostics
- Astronomical spectroscopy and imaging, including remote detection of X-rays
- Synchrotron diagnostics and testing of beamline optical components
- Measurement and verification of X-ray optical data

<u>Scattering</u>	05. Hard X-Ray Diffraction
	06. Macromolecular Crystallography
	07. Hard X-Ray Scattering
	08. Soft X-Ray Scattering

Scattering: 05. Hard X-Ray Diffraction

How it works:

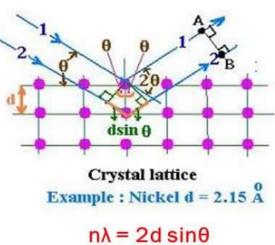
X-rays are scattered (diffracted) into very specific directions when illuminating a crystalline sample

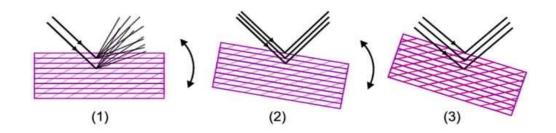
BRAGG'S LAW

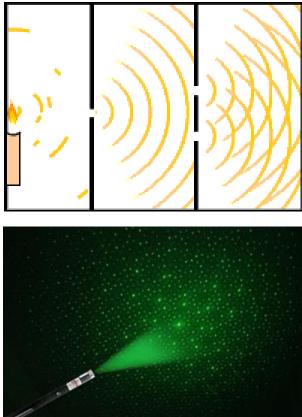
Angle of incident ray = angle of diffracted ray

$n\lambda = 2dsin\theta$

- 2θ = angle between incident and reflected beams
- d = spacing between planes
- λ = wavelength
- n = order of diffraction







Scattering: 05. Hard X-Ray Diffraction

Purpose:

Determination of the atomic structure of materials

Typical applications:

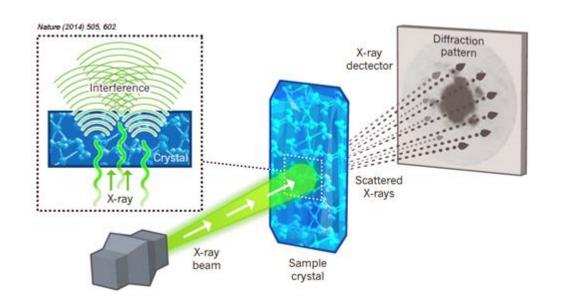
- Structural studies of crystalline materials
- Drug design by pharmaceutical industry
- Biomineralization
- New microporous materials including natrolites, phosphates, and titanates
- Novel complex oxides: structure-property relationships, phase transitions
- Residual stress determination in situ

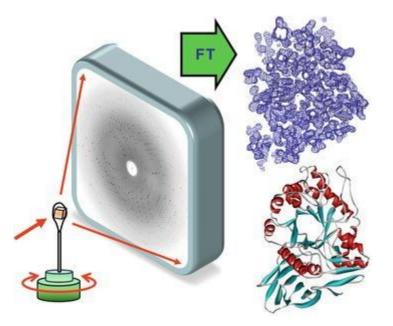
- X-ray diffraction (XRD):
 - powder, single-crystal, and surface diffraction
- X-ray standing wave (XSW)

Scattering: 06. Macromolecular Crystallography

How it works:

X-rays are scattered (diffracted) into very specific directions when illuminating a crystalline sample





Scattering: 06. Macromolecular Crystallography

Purpose:

Determination of the three-dimensional structure of large biological molecules

Typical applications:

- Therapeutic drug design
- Enzyme mechanisms
- Supramolecular structure
- Molecular recognition
- Nucleic acids
- Structural genomics
- High-throughput crystallography

Includes:

• Protein crystallography (PX)

Scattering: 07. Hard X-Ray Scattering

How it works:

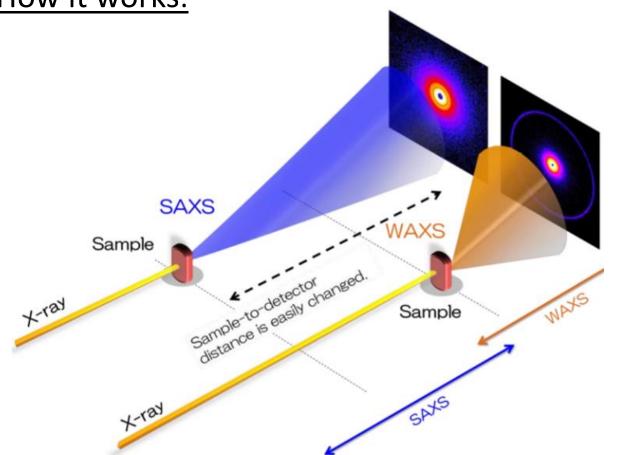
When a sample is illuminated by x-rays, these incident x-rays can be deflected and scattered by the sample, producing complex patterns. Analysis of these patterns, their intensities as well as the angle of scatter (incident vs scattered x-rays), changes in polarization, wavelength, and/or energy, can reveal structural, elemental and atomic information about the sample. There are two principal variants:

- Elastic: the energy (wavelength) of the detected X-ray is the same as that of the incident X-ray
- Inelastic: the energy of the detected X-ray is lower than that of the incident Xray. The lost of energy is carried away by a vibrational, electronic, or magnetic excitation. The detection system in this case requires a spectrometer to measure the energy loss

https://www.princetoninstruments.com/learn/x-ray-scattering/intro-to-x-ray-scattering Experimental Techniques at Light-Source Beamlines – U.S. Synchrotron Radiation Light Sources

Scattering: 07. Hard X-Ray Scattering

How it works:





https://www.princetoninstruments.com/learn/x-ray-scattering/intro-to-x-ray-scattering Courtesy of Marc Malfois (SAXS Workshop 2014, ALBA)

Scattering: 07. Hard X-Ray Scattering

Purpose:

Measures the patterns, intensities, and angles of scatter produced when a sample is illuminated by a monochromatic X-ray beam. Information on shape, dispersity, sample size, porosity, morphology, and orientation are some of the few things that can be measured via X-ray scattering

Typical applications:

- Liquid-vapor, liquid-liquid, and molecular film interfaces
- Colloids, solution-phase protein, polymers
- Collective dynamics in soft materials
- Short-range order in amorphous materials
- Phonons and elementary excitation in solids
- Electron momentum distribution in solids

Scattering: 07. Hard X-Ray Scattering

- Small-angle X-ray scattering (SAXS)
- Wide-angle X-ray scattering (WAXS)
- Grazing-incidence small-angle X-ray scattering (GISAXS)
- X-ray Raman scattering
- Compton scattering
- Inelastic X-ray scattering (IXS)
- Resonant inelastic X-ray scattering (RIXS)
- Nuclear resonant scattering (NRS)
- X-ray photon correlation spectroscopy (XPCS)

<u>Scattering</u>: 08. Soft X-Ray Scattering

How it works:

It is a photon-in/photon-out technique.

Purpose:

Soft x-ray scattering techniques employ the excitation of electrons in relatively shallow core energy levels (100–2000 eV) to probe the electronic structure and other properties of various kinds of matter

Typical applications:

- Strongly correlated materials
- Magnetic materials
- Environmental science
- Catalysis

Scattering: 08. Soft X-Ray Scattering

- Soft X-ray emission spectroscopy (SXES)
- Inelastic X-ray scattering (IXS)
- Resonant X-ray inelastic scattering (RIXS)
- Speckle patterns
- Small-angle X-ray scattering (SAXS)

<u>Imaging</u>	09. Hard X-Ray Imaging 10. Soft X-Ray Imaging 11. Infrared Imaging 12. Lithography
	12. Lithography

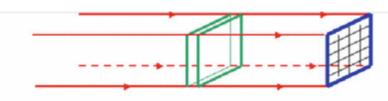
Imaging: 09. Hard X-Ray Imaging

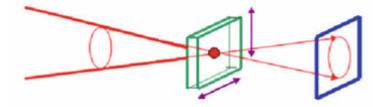
How it works:

There are two basic experimental methods:

- Imaging: a full-field image can be recorded with a parallel beam with a pixelated detector (each pixel correspond to a given volume in the sample).
 The spatial resolution of the image is associated with the pixel size.
- Scanning: a very small illuminated spot is created on the sample using focusing devices. The image is then built up by "raster scanning" the sample through the illuminated spot

In either case, there has to be a physical cause of the contrast. This can be due to changes in absorption, elemental composition, or refractive index of the sample.





Imaging: 09. Hard X-Ray Imaging

Typical applications:

- Human and animal physiology
- Mapping of magnetic domains in two dimensions
- Mapping of composite materials in three dimensions (tomography)
- Properties of individual grains in a polycrystalline material
- Mapping of the distribution of elements in cells
- Strains in near-perfect crystals
- Time-resolved imaging of sprays

- Radiography
- Phase contrast imaging
- Scanning micro/nanoprobe
- Fullfield microscopy

- Diffraction enhanced imaging (DEI)
- X-ray tomography
- Topography

Imaging: 10. Soft X-Ray Imaging

How it works:

Employs photons or electron optics to achieve the necessary high resolution

Purpose:

The wavelengths of soft X-ray photons (1-15 nm) are very well suited to probe the interior structure of biological cells and inorganic mesoscopic systems. For example, cell biology uses radiation around 300-500 eV, while nanomagnetic studies need 600-900 eV

Typical applications:

- Cell biology
- Nanomagnetism
- Environmental science
- Soft matter, polymers

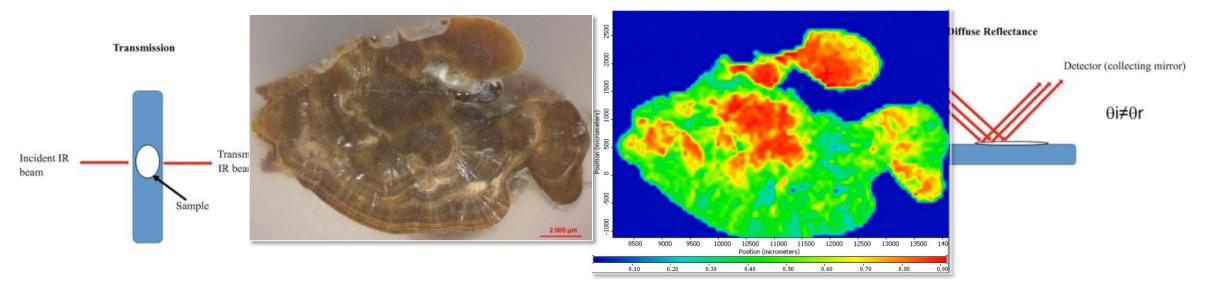
Imaging: 10. Soft X-Ray Imaging

- Soft X-ray imaging
- Photoelectron emission microscopy (PEEM)
- Scanning transmission X-ray microscopy (STXM)
- Full-field microscopy
- X-ray diffraction imaging (XDI)
- X-ray tomography
- Computer-aided tomography (CAT) scans

Imaging: 11. Infrared Imaging

How it works:

IR synchrotron radiation is focused through, or reflected from, a small spot on the specimen and then analyzed using a spectrometer. Tuning to characteristic vibrational frequencies (obtained in the analyzed spectra) serves as a sensitive fingerprint for molecular species. Then, images of the various species are built up by raster scanning the specimen through the small illuminated spot.



Abidi, N. (2021). Introduction to FTIR Microspectroscopy. In: FTIR Microspectroscopy . Springer, Cham. https://doi.org/10.1007/978-3-030-84426-4_1

Imaging: 11. Infrared Imaging

Typical applications:

- Chemistry in biological tissues
- Chemical identification and molecular conformation
- Environmental biodegradation
- Mineral phases in geological and astronomical specimens
- Electronic properties of novel materials
- Forensic studies

- Infrared (IR)
- Microspectroscopy
- Infrared imaging
- Infrared microprobe

Imaging: 12. Lithography

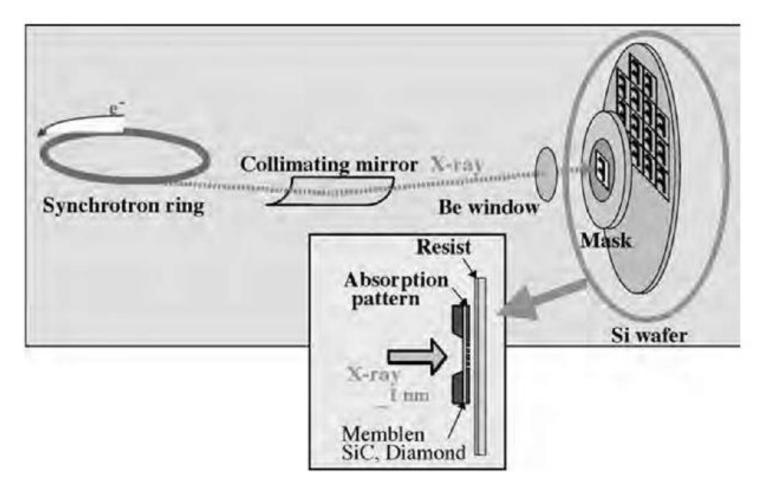
How it works:

X-ray lithography (XRL) is an advanced version of optical lithography in which shorter wavelengths are used. A special type of mask is used with different local X-ray absorption areas to define the patter. Then, this pattern is replicated on an X-ray sensitive material called a resist, which was previously deposited on a substrate. When the X-ray passing through the pattern falls on the resist, it may cause crosslinking (for negative resists) or bond breaking (for positive resists), depending on the chemical nature of the resist. After exposure, the whole thing is dipped in a specific solvent to reveal the patter. This is how X-ray lithography creates nanopatterns on the substrate

33

Imaging: 12. Lithography

How it works:

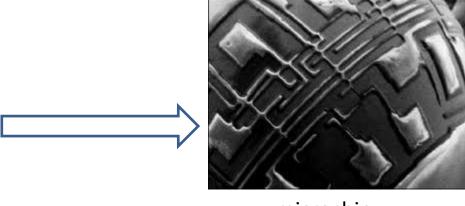


Imaging: 12. Lithography

Includes:

- Extreme ultraviolet (EUV) lithography
- Interferometry
- LIGA (lithography, electroplating and molding)
- Microelectromechanical structures (MEMS)
- X-ray lithography
- Nanolithography





microchip

TUNTWIN's Workshop



Merci! Thank you! ¡Gracias!



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