TUNTWIN's Workshop

Session C: Scattering and diffraction techniques session







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Session: Spectroscopy techniques

Synchrotron based X-ray Diffraction

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Synchrotron based X-ray diffraction

➤Introduction

> Why use synchrotron radiation?

> Examples

Synchrotron based X-ray diffraction

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What is diffraction?

X-rays scattered by a set atoms produce X-ray radiation in all directions, leading to interferences due to the coherent phase differences between the interatomic vectors that describe the relative position of atoms.



Bending of waves around the corners of an obstacle or through an aperture into the region of geometrical shadow of the obstacle/aperture

How it works? Bragg's Law

"Crystals, at certain specific wavelengths and incident angles, produce intense peaks of reflected radiation (known as Bragg peaks)" W. L. Bragg







- 2θ = angle between incident and reflected beams
- d = spacing between planes
- λ = wavelength
- n = order of diffraction



How it works? Bragg's Law

Each Bragg reflection (hkl) is associated to a plane of atoms as described by crystallography.





Single crystal X-ray diffraction



Powder X-ray diffraction





Powder X-ray diffraction

- \blacktriangleright Peak position $2\theta_{hkl}$:
 - information on overall periodical arrangement of atoms
- Peak intensity I_{hkl}:
 - Type of atom in material
 - Position (x, y, z) of atoms in the structure (unit cell)
- Peak FWHM_{hkl}:
 - Microstructural features
 - Strains Ο
 - Size \bigcirc
 - Stacking faults Ο
 - Ο ...





rings

Strained

Powder X-ray diffraction

Typical applications:

- Qualitative and quantitative phase analysis
- Structural analysis
- Stability and phase transitions
- Microstructure (crystalline size, microstrains)
- Microdiffraction

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Synchrotron based X-ray diffraction: Why use synchrotron?

Main Characteristics

- ➢ High intensity
- > High monochromaticity
- Wavelength can be varied between experiments

Advantages

- High signal-to-noise ratio
- High angular resolution
- High speed of data collection



Synchrotron based X-ray diffraction: Why use synchrotron?

Sample environments

- FMB Oxford hot air blower
- > Oxford cryostream 700 series
- > Dynaflow liquid He cryostat
- Capillary flow cell
- Equipment for the electrochemical studies
- \succ On-line pressure calibration set-up (ruby lumiscence method)
- BETSA external ring heater
- External heating vacuum system
- IHE cryostat
- Navitar 12X online visualization system



















Synchrotron based X-ray diffraction

>Introduction

> Why use synchrotron radiation?



1. X-ray Total Scattering Study of Phases Formed from Cement Phases Carbonation







Figure 3. Selected low angle ranges (intensity vs 20) of the synchrotron Rietveld plots ($\lambda = 0.41 \text{ Å}$) for (**a**) Y-A_hyd, hydrated for 1.5 months and (**b**) Y-A_carb paste after carbonation for 80 days at 3% of CO₂ (T = 20 °C and RH = 65%). The top pattern contains crystalline ettringite and very broad diffraction peaks of nano-gibbsite. The main peaks are labelled. Top pattern: ettringite (E) and nanocrystalline-gibbsite (**nc-G**). Bottom pattern: aragonite (A), bassanite (B), nanocrystalline-gibbsite (**nc-G**) [including their (hkl) indexes].

2. Operando Synchrotron X-ray Diffraction in Calcium Batteries: Insights into the Redox Activity of 1D Ca₃CoMO₆ (M = Co and Mn)



Figure 3. Sample holder with four coin cells to be simultaneously monitored operando, single coin-cell image, and an exploded view of a coin cell.

<u>2. Operando Synchrotron X-ray Diffraction in Calcium Batteries: Insights into the Redox</u> <u>Activity of 1D Ca₃CoMO₆ (M = Co and Mn)</u>



Figure 4. Potential vs capacity profiles from GCPL experiments on $Ca_3Co_2O_6//AC$ cells using $Ca(BF_4)_2$ in EC/PC as the electrolyte at RT and C/50 rate (a), corresponding SXRD patterns (b) and zoomin images (c,d), selected patterns correspond to the first and last pattern of cell (A) in black and light red, last oxidation and last reduction of cell (B) in red and dark green, and last ex situ pattern cell (B) in blue. (e,f) Zoom-in images of reflections 100 and 300 of in situ SXRD patterns of cells (A,B), respectively.





<u>3. Structural evolution, optical gap and thermoelectric properties of CH₃NH₃SnBr₃ hybrid perovskite, prepared by mechanochemistry</u>



Fig. 2 (a) Thermal evolution of the (100) and (220) cubic lines from synchrotron diffraction. (b) Observed (circles), calculated (full line) and difference (bottom) Rietveld profiles for MASnBr₃ for SXRD data at 160 K. (c) Final crystal structure after Rietveld refinement at 160 K. (d) Evolution of the unit-cell parameters at different temperatures. The upper colour bars correspond to the calorimetric regions I, II and III.

C. A. Lopez, et al. Mater. Adv., 2021, 2, 3620. DOI: 10.1039/d1ma00196e

<u>4. Determination of the Crystal Structures in the A-Site-Ordered YBaMn₂O₆ Perovskite</u>



Figure 2. Details of the SXRPD patterns showing (a) the changes of the splitting in main diffraction peaks, indicating a unit cell change, and (b) the occurrence of different superstructure peaks at different temperatures. The asterisk marks the main peak of the secondary phase Y_2O_3 (0.3% in weight). (c) Temperature dependence of selected superstructure peaks. The subscript T in all panels indicates that indexation of the peaks is related to the parent tetragonal cell.

5. Crystal Structure of BaCa(CO₃)₂ Alstonite Carbonate and Its Phase Stability upon Compression



Figure 4. Powder XRD patterns of $BaCa(CO_3)_2$ alstonite at different pressures. The pattern of the recovered alstonite sample is shown on top. Asterisks denote the diffraction maxima of copper, the internal pressure gauge.

R. Chulia-Jordan, et al. ACS Earth Space Chem. 2021, 5, 5, 1130–1139. doi.org/10.1021/acsearthspacechem.1c00032

<u>6. Unveiling the Structural Behavior under Pressure of Filled $M_{0.5}Co_4Sb_{12}$ (M = K, Sr, La, Ce, and Yb) Thermoelectric Skutterudites</u>



Figure 3. Synchrotron X-ray diffraction patterns of $\Box Co_4Sb_{12}$ under 7.0 GPa (a) and 11.9 GPa (c). Details on the pressure evolution of Sb metal secondary phase (b), showing its phase transition from a rhombohedral (S.G. $R\overline{3}m$) to close-packed hexagonal (S.G. $P6_3/mmc$) phase around 9.4 GPa.

7. Discriminating the origin of calcium oxalate monohydrate formation in kidney stones

via synchrotron microdiffraction



Fig. 4 2D-µXRD images of four representative points from calcium oxalate kidney stones, namely COM (top-left), COD (top-right), TRA (bottom-left) and MIX (bottom-right).

Fig. 5 Representation of the azimuthal plots for three samples, one representative of COM (top, S23), one representative of TRA (middle, S26A) and a test sample (bottom, S33), at reflections 100 and 040, left and right panels, respectively. The orange square (bottom of each plot) represents the resin where the stones are embedded, and the red dotted lines represent the two regions characterized as COM. The separation between points is 300 a.u. on the *y*-axis.





8. Synchrotron X-ray microdiffraction to study dental structures in Cretaceous crocodylomorphs



Fig. 3. Two-dimensional X-ray Diffraction (2D-XRD) patterns of the fossil tooth G2-W-016 (A), in the enamel (left) and the dentine (right). The enamel point is located at 70 µm and the dentine at 200 µm from the surface. Powder diffraction patterns obtained from the 2D-XRD data of the same tooth (B). 2D-XRD patterns of the modern tooth KD6 (C), in the enamel (left) and the dentine (right). The enamel point is located at 120 µm and the dentine at 600 µm from the surface. Powder diffraction patterns obtained from the 2D-XRD data of the same tooth (D).

O. Vallcorba, et al. Cretaceous Research 128 (2021) 104960. doi.org/10.1016/j.cretres.2021.104960

8. Synchrotron X-ray microdiffraction to study dental structures in Cretaceous crocodylomorphs Fossil Modern



Fig. 4. Evolution of the hydroxyapatite average crystallite size in the *a/b* (*h*k0) and *c* (001) directions with the teeth depth, for the fossil G2-W-016 (A) and the modern KD6 (B) teeth Av: Average, Scale bar = 100 µm.



O. Vallcorba, et al. Cretaceous Research 128 (2021) 104960. doi.org/10.1016/j.cretres.2021.104960 **Fig. 5.** Azimuthal plots from two-dimensional X-ray Diffraction (2D-XRD) for a single enamel point of the fossil G2-W-016 (A) and the modern KD6 (B) teeth. The hkl indices of the reflections are written on the top of each plot and the colour scale corresponds to the reflection intensities (detector counts). Stacking of azimuthal plots for the (002) reflection at each measured point of the same fossil (C) and modern (D) samples. Relative intensity at the same colour scale as the above plots. The preferred orientations of the *c* axis direction of the hydroxyapatite crystal structure for both teeth are represented on the respective images. Scale bar = 100 μm.

9. The color of the circus mosaic from Barcino (Roman Barcelona): Characterization, provenance and technology issues



Figure 1. In the center, a general view of the Circus mosaic, and the smaller surrounding images illustrate the zoomed parts to identify the numbered tesserae in their original locations before their extraction.

Figure 11. Representative diffractograms obtained by SR tts-µXRD from the crystalline features found within the green glass tessera #12: (a) POM image in RL mode with several highlighted colored rectangles, also shown as insets in the diffraction patterns, where the corresponding irradiated areas are marked with circles: ((**right**) grayish reflectance crystals including malayaite (mly) and cassiterite (cst), (**bottom-left**) pore filling including cerussite (cer) and calcite (cal); (**bottom-right**) reddish relict clod including quartz (qz), hematite (hm), diopside (dio) and cordierite (crd)). (b) POM image in RL mode with a highlighted rectangle, also shown as an inset (along with the equivalent SEM image) in the diffraction pattern on the right, where the corresponding irradiated area is marked with a circle, they correspond to yellowish reflectance crystals of cubic PbSnO₃.



L. Casas, et al. Minerals 2021, 11(7), 746; doi.org/10.3390/min11070746

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Merci! Thank you! ¡Gracias!



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